Benzotriazole-Assisted Preparations of 2-(Substituted amino)pyridines and Pyrid-2-ones

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Base-promoted reactions of benzotriazolyl-containing acetic acid derivatives, 2-(benzotriazol-1-yl)acetonitrile (**7a**), 2-(benzotriazol-1-yl)acetamide (**7b**), and (\pm)-2-(benzotriazol-1-yl)propionamide (**7c**), with α , β -unsaturated ketones **8** give efficient and regioselective access to previously difficult to attain *3-unsubstituted* pyridine derivatives: the 2-(substituted amino)pyridines **14a**–**k** and the 4,6-substituted pyrid-2-ones **15a**–**h**. The pyridine rings result from tandem [3 + 3] annulations involving a Michael addition followed by cyclization.

Introduction

2-(Substituted amino)pyridines were previously mostly prepared (i) by reaction of primary or secondary aliphatic amines with 2-halopyridines $^{1-5}$ or with imidol silvl ethers derived from the corresponding pyrid-2-ones,⁶ (ii) by aminolysis of 2-alkoxypyridines,³ and (iii) by pyridine ring formation reactions, usually of the [4 + 2] or [3 + 3]type.^{7,8} Of these methods, pathway i is the most concise and efficient but requires high temperatures and pressure.^{1,3,5} Method ii found application only for alkoxypyridines containing activating groups in the pyridine ring.³ Ring formation reactions iii in the aminopyridine series are quite rare and have found limited application.⁸ A previous report from our laboratory disclosed efficient N-alkylations of 2-aminopyridines, achieved by NaBH₄ reduction of the adducts formed from benzotriazole, aldehydes, and 2-aminopyridines.9

Many syntheses of pyridones^{7,8} involve a CCN fragment of cyanoacetamide or malononitrile in a [3 + 3]annulation reaction, thus leaving a cyano group in the C(3)-position of the pyridone system.⁸ Such reactions are regioselective and involve a Michael addition of a malonic acid derivative to an α,β -unsaturated carbonyl compound, followed by cyclization and aromatization of an intermediate dihydropyridone. Syntheses of *C(3)-unsubstituted* pyridones are less common: recently, Barluenga *et al.* utilized the ring closure of acylated aminoazabutadienes in the presence of LDA.¹⁰

We now report that reactions of α , β -unsaturated ketones with 2-(benzotriazol-1-yl)acetonitrile and *sec*-

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amines in ethanol readily afford the corresponding 2-(substituted amino)pyridines. Furthermore, 2-(benzo-triazol-1-yl)acetamide and (\pm) -2-(benzotriazol-1-yl)propionamide under similar conditions, but in the presence of NaOH, smoothly form the corresponding substituted pyrid-2-ones.

Previously, the reaction of α , β -unsaturated carbonyl compounds and malononitrile in the presence of piperidine as a base was shown¹¹ to give 2-amino-3-cyano-2*H*-pyrans **2** as major reaction products (Scheme 1), whereas when NaOH was used as a base, 2-alkoxy-3-cyanopyridines **3** were isolated directly. Pyrans **2** were subsequently converted into 2-alkoxypyridines **3** by sodium alkoxides. Recently, Ciufolini and co-workers prepared substituted 3-cyanopyrid-2-ones **5** from various enones and enals with cyanoacetamide in DMSO using *t*-BuOK as a base under an oxygen atmosphere.¹² Their proposed mechanism involves a molecule of oxygen in the single electron transfer (SET) process, thus enabling the aromatization of the ring.

We reasoned that $\hat{2}$ -alkoxypyridines **3** were probably the key intermediates in both pathways utilizing either malononitrile or cyanoacetamide. Thus, the formation

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Table 1. Preparation of Benzotriazole-Substituted **CH-Active Intermediates 7**

| compd | | X | yield, % | mp, °C (solvt) | molecular formula | elemental analysis, found (required) | | |
|------------|----|-------------------|-------------|--|---|---|----------------|------------------|
| 7 | R | | | | | С | Н | Ν |
| a | Н | CN | 72 | 87–88 (85–86) ^a (benzene) | $C_8H_6N_4$ | | | |
| b | Н | CONH ₂ | 81 | 163–164 (EtOAc) | C ₈ H ₈ N ₄ O | 54.85 (54.54) | 4.51 (4.58) | 31.68 (31.80) |
| С а і ; | Me | CONH ₂ | 67 | 142 (MeOH) | C ₉ H ₁₀ N ₄ O | 56.65 (56.83) | 5.30 (5.30) | 29.42 (29.46) |

of pyran system 2 is followed by "base-catalyzed ring opening and recyclization" to form 2-alkoxypyridines 3," ¹¹ by analogy with pyrylium/pyridinium conversion.¹³ Although transformation of pyridines 3 into pyridones 5 was not yet established, Ciufolini et al.¹² proposed that malonamic ester EtOOCCH₂CONH₂ reacts with 1-(1propyl)-3-(quinolin-2-yl)prop-1-en-3-one in DMSO/t-BuOK with the formation of an intermediate anion whose structure was similar to 4 (Scheme 1). The presence of a labile benzotriazole moiety in place of the cyano group in structure 1 should facilitate the formation of the adjacent double bond and thus facilitate the formation of either pyridine 3 or pyridone 5.

Results and Discussion

We employed a new method for the preparations of previously described 2-(benzotriazol-1-yl)acetonitrile (7a) and 2-(benzotriazol-1-yl)acetamide (7b) and also of the new (\pm) -2-(benzo-triazol-1-yl)propionamide (7c). Compound 7a was previously synthesized by reaction of 1-(chloromethyl)benzotriazole with sodium cyanide in DMSO for 3 days¹⁴ and was subsequently converted into **7b** by treatment with H_2O_2 in DMSO.¹⁵ We prepared compounds $7\mathbf{a} - \mathbf{c}$ in good yields by reaction of sodium benzotriazolate with the corresponding halides 6a-c in refluxing toluene in the presence of a phase-transfer catalyst (18-crown-6) (Scheme 2). Compounds 7a-c (Table 1) were characterized by elemental analysis and NMR.

Cyano derivative 7a reacted with 1,3-diphenylprop-2en-1-one (8a) in the presence of diethylamine in ethanol to give 2-(diethylamino)-4,6-diphenylpyridine (14a) (64%) (Scheme 3). 2-(Disubstituted amino)pyridines 14b-k (Table 2) with diverse substituents at the C(4) and C(6) positions of the pyridine ring were similarly synthesized in yields of 35-74%. In the series of 2-(disubstituted amino)pyridines 14 prepared, those containing a methyl



substituent at the C(6) position, *i.e.*, **14i-k**, were obtained in lower yield in comparison to their C(6)-arylsubstituted analogs (14a-h). An extra aryl group evidently facilitates cyclization of intermediates 9.

Reaction of cyano derivative 7a with chalcone 8a in ethanolic NaOH for 18 h at 20 °C gave after column chromatography 4,6-diphenylpyrid-2-one (15a, 47%) and 2-ethoxy-4,6-diphenylpyridine (13a, 15%) (Scheme 3). Compound 13a, with a structure analogous to that of 3 (Scheme 1), is a probable intermediate that is hydrolyzed to pyrid-2-one 15a. Alternatively, 15a could be formed by direct nucleophilic attack of hydroxide anion rather than ethoxide anion as shown in Scheme 3. The analogous reaction of 4-phenylbut-3-en-2-one (8h) with 2-(benzotriazol-2-yl)acetonitrile (7a) gave only a low yield of 2-ethoxy-4-phenyl-6-methylpyridine (13i).

1,3-Disubstituted prop-2-en-1-ones 8a-g reacted with 2-(benzotriazol-1-yl)acetamide (7b) to form the corresponding 4-phenyl-6-(substituted)pyrid-2-ones 15a-h (53-92%) (Scheme 4, Table 3). A substituent in the α -position of the α,β -unsaturated ketones prevents formation of detectable amounts of pyrid-2-ones: attempts to use either 1,3-diphenyl-2-methylprop-2-en-1-one or 2-(benzylidene)cyclohexan-1-one failed. When 2-butyl- or 2-benzyl-2-(benzotriazol-1-yl)acetamides were employed in the reaction with chalcone 8a, the corresponding 3-butyl- or 3-benzyl-4,6-diphenylpyrid-2-ones were detected in the reaction mixtures by GC/MS in less than 10% yields. The bulkier the substituent, the more difficult the formation of intermediate 16.

The structures of compounds 15 were supported by ¹H and ¹³C NMR spectra, GC/MS, and elemental analyses

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Table 2. Preparation of 2-(Disubstituted amino)pyridines 14

| | | | | | | | elemental analysis found (require | | |
|----------|-----------------------------------|----------------|----------------|----------|-----------------|--------------------|-----------------------------------|--------|-------------|
| compd 14 | R | \mathbb{R}^1 | \mathbb{R}^2 | yield, % | mp, °C (solvt) | molecular formula | С | Н | Ν |
| а | Ph | Et | Et | 64 | 112-113 | $C_{21}H_{22}N_2$ | 83.23 | 7.46 | 9.22 |
| | DI | 10 | TT \ | ~ 4 | (hexanes) | | (83.40) | (7.33) | (9.26) |
| D | Ph | -(C | $H_2)_4 -$ | 74 | 161-162 | $C_{21}H_{20}N_2$ | 83.71 | 6.83 | 9.14 |
| | DI | (0) | | | (hexanes) | | (83.96) | (6.71) | (9.33) |
| С | Ph | $-(CH_2)_5-$ | | 68 | 130-131 | $C_{22}H_{22}N_2$ | 84.31 | 7.32 | 8.81 |
| | 51 | (011) | 0 (011) | ~ ~ | (hexanes) | a | (84.04) | (7.05) | (8.91) |
| d | Ph | $-(CH_2)_2$ | $O(CH_2)_2 -$ | 65 | 153-154 | $C_{21}H_{20}N_2O$ | 79.67 | 6.42 | 8.83 |
| | | | | | (hexanes/EtOAc) | | (79.72) | (6.37) | (8.85) |
| е | $4 - MeC_6H_4$ | Et | Et | 66 | 99-100 | $C_{22}H_{24}N_2$ | 83.52 | 7.78 | 8.87 |
| | | | | | (EtOH) | | (83.50) | (7.64) | (8.85) |
| f | 4-MeC ₆ H ₄ | -(C | $H_2)_4 -$ | 70 | 176 - 177 | $C_{22}H_{22}N_2$ | 83.76 | 7.08 | 8.86 |
| | | | | | (EtOH) | | (84.04) | (7.05) | (8.91) |
| g | 4-MeC ₆ H ₄ | -(C | $(H_2)_5 -$ | 64 | 134 - 135 | $C_{23}H_{24}N_2$ | 84.10 | 7.49 | 8.50 |
| | | | | | (EtOH) | | (84.11) | (7.36) | (8.53) |
| h | 4-MeC ₆ H ₄ | $-(CH_2)_2$ | $O(CH_2)_2 -$ | 56 | 153 - 154 | $C_{22}H_{22}N_2O$ | 80.13 | 6.77 | 8.47 |
| | | | | | (EtOAc) | | (79.97) | (6.71) | (8.48) |
| i | Me | Et | Et | 43 | oil | $C_{16}H_{20}N_2$ | | | 11.55^{a} |
| | | | | | | | | | (11.66) |
| i | Me | -(C | $(H_2)_5 -$ | 35 | oil | $C_{17}H_{20}N_2$ | 80.78 | 8.18 | 10.97 |
| 0 | | | | | | | (80.91) | (7.99) | (11.10) |
| k | Me | $-(CH_2)_2$ | $O(CH_2)_2 -$ | 45 | 116-117 | $C_{16}H_{18}N_2O$ | 75.41 | 7.20 | 10.97 |
| | | . 107 1 | | | (hexanes) | | (75.56) | (7.13) | (11.01) |

^a MS: found 241.1701 [M + 1] calcd 241.1705 [M + 1].





and also by the identity of the melting point of compound 15a samples obtained according to the methods depicted in both Schemes 3 and 4.

A suggested mechanism for the formation of both 2-aminopyridines and pyrid-2-ones by the reaction of nitrile 7a with chalcones 8 in the presence of bases (Scheme 3) involves intermediate Michael adducts 9 and 10, which upon nucleophilic attack by either a *sec*-amine or an alcohol form the corresponding tetrahydropyridines 11 and 12. Aromatization of the reaction intermediates occurs upon dehydration and loss of the benzotriazole molecule leading to the appropriate pyridines 14 and 13. In the case of 2-ethoxypyridine 13, the excess of a strong base promotes the loss of the alkoxy moiety and subsequent tautomerization of 2-hydroxypyridine into pyridone 15.

Reaction of benzotriazole-containing amides 7b,c with 1,3-disubstituted prop-2-en-1-ones 8 (Scheme 4) also involves analogous tandem reaction sequence. However, here the pyridone backbone is formed immediately after cyclization of Michael adduct into piperidin-2-one 16, which undergoes dehydration and loss of benzotriazole to give the desired pyrid-2-ones 15a-h.

In summary, CH-active benzotriazolyl derivatives of acetic acid **7a**–**c** react with α , β -unsaturated ketones to provide an efficient one-pot method for the regioselective preparation of both 2-(substituted amino)pyridines 14 and 3,4,6-substituted pyrid-2-ones 15. This gives access to the 3-unsubstituted pyridine derivatives, which were previously difficult to obtain. The formation of the heterocycle is the result of a [3 + 3] annulation reaction involving the addition of the CH-active compound to the double bond, followed by heterocyclization. The aromatization of the heterocycle thus formed is further assisted by the loss of the benzotriazole moiety, which enhances the synthetic applicability of our new methodology.

Experimental Section

General Methods. See refs^{17,18} Substituted prop-2-en-1ones 8b,c,g were synthesized in accordance with literature procedure;¹⁹ others were commercially available and were used as received. ¹H NMR spectra were recorded at 300 MHz in CDCl₃, except for compound 7c (DMSO-d₆) and ¹³C NMR spectra at 75 MHz in the same solvents, except for compounds **15d-f**, whose spectra were recorded in CDCl₃ containing 5% (v/v) of TFA. The ¹³C NMR spectra of pyridones **15a**-h consist of a number of overlapped signals. The peak values are reported as observed.

2-(Benzotriazol-1-yl)acetonitrile (7a). A mixture of sodium benzotriazolate (8.46 g, 60 mmol), 2-chloroacetonitrile (4.53 g, 60 mmol), and 18-crown-6 (50 mg) in toluene (120 mL) was stirred at reflux for 18 h. On cooling, the mixture was washed with water, the organic layer was separated and dried over anhyd Na₂SO₄, and the solvent was removed in vacuo to give a crude product, which was recrystallized from benzene.

2-(Benzotriazol-1-yl)acetamide (7b). The product was prepared as described above from sodium benzotriazolate (2.54 g, 18 mmol), bromoacetamide (2.48 g, 18 mmol), and 18crown-6 (25 mg) in refluxing toluene (60 mL). On cooling, the mixture was poured onto water (100 mL) and thoroughly stirred and the precipitate collected by filtration. The precipitate was washed with benzene and water and dried in vacuo to give the product as white needles. The analytical sample was obtained by recrystallization from ethyl acetate: ¹H NMR δ 5.43 (s, 2H), 7.40 (t, J = 7.5 Hz, 1H), 7.46 (br s, 1H), 7.55 (t, J = 7.6 Hz, 1H), 7.77 (d, J = 8.4 Hz, 1H), 7.86 (br s, 1H), 8.04 (d, J = 8.3 Hz, 1H); ¹³C NMR δ 49.7, 110.9, 119.0, 123.8, 127.2, 133.8, 145.1, 167.5.

(±)-2-(Benzotriazol-1-yl)propionamide (7c). A mixture of sodium benzotriazolate (1.30 g, 9.2 mmol), (±)-2-bromopropionamide (6, X = Br, R = Me) (0.97 g, 6.4 mmol), and 18crown-6 (50 mg) in toluene (30 mL) was stirred under reflux for 18 h. On cooling, water was added to the reaction mixture, and the precipitated crude product was collected by filtration, washed with water and toluene, and air dried. Recrystallization from methanol gave the benzotriazol-1-yl isomer of 7c as a white solid (0.81 g, 67%): ¹H NMR δ 1.88 (d, J = 7.1 Hz,

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| Table 3. | Preparation | of 3,4,6-Substituted | l Pyrid-2-ones 15 |
|----------|-------------|----------------------|-------------------|
| | | , , | ./ |

| | | | | | | | elemental analysis found (required) | | |
|-------------|------------------------------------|-----------------------------------|----------------|----------|--|--------------------------------------|-------------------------------------|----------------|-----------------------------|
| pyridone 15 | R | \mathbb{R}^1 | \mathbb{R}^2 | yield, % | mp, °C (solvt) | molecular formula | С | Н | Ν |
| а | C ₆ H ₅ | C ₆ H ₅ | Н | 92 | 216–217 (205–206) ^a (EtOAc) | C ₁₇ H ₁₃ NO | | | |
| b | 4-MeC ₆ H ₄ | C_6H_5 | Н | 85 | 238–239 (MeOH) | C ₁₈ H ₁₅ NO | 82.44 (82.73) | 5.67 (5.79) | 5.31 (5.36) |
| С | 4-MeOC ₆ H ₄ | 4-MeC ₆ H ₄ | Η | 54 | 258–260 (EtOAc) | $C_{19}H_{17}NO_2$ | 78.04 (78.33) | 5.96 (5.88) | 4.61 (4.81) |
| d | C_6H_5 | $4-O_2NC_6H_4$ | Η | 57 | 281–282 (acetone) | $C_{17}H_{12}N_2O_3$ | (69.86) | (4.14) | 9.56 ^b (9.58) |
| e | C_6H_5 | 4-ClC ₆ H ₄ | Η | 53 | 237–238 (acetone) | C ₁₇ H ₁₂ ClNO | 72.35 (72.47) | 4.23 (4.29) | 4.96 (4.97) |
| f | 4-ClC ₆ H ₄ | 4-ClC ₆ H ₄ | Η | 75 | 308–309 (acetone) | $C_{17}H_{11}Cl_2NO$ | 64.66 (64.58) | 3.50 (3.51) | 4.38 (4.43) |
| g | 1-naphthyl | $3-O_2NC_6H_4$ | Η | 31 | 218–220 (EtOAc) | $C_{21}H_{14}N_2O_3$ | (73.68) | (4.12) | с (7.86) |
| h | C ₆ H ₅ | C ₆ H ₅ | Me | 53 | 248–249 (CHCl ₃) | C ₁₈ H ₁₅ NO | 82.48 (82.73) | 5.94 (5.79) | 5.35 (8.18) |

^a Lit.¹⁶ mp. ^b MS: found 292.0838 [M⁺], calcd 292.0848 [M⁺]. ^c MS: found 343.1076 [M + 1], calcd 343.1083 [M + 1].

3H), 5.67 (q, J =7.1 Hz, 1H), 7.36–7.48 (m, 2H), 7.53 (t, J = 7.1 Hz, 1H), 7.75–7.85 (m, 2H), 8.04 (d, J = 8.3 Hz, 1H); ¹³C NMR δ 16.8, 57.4, 111.2, 119.1, 123.9, 127.1, 132.7, 145.5, 170.3.

General Procedure for the Preparation of 2-(Disubstituted amino)pyridines 14a–k. To a stirred solution of 1-(cyanomethyl)benzotriazole (**7a**) (0.395 g, 2.5 mmol) and the corresponding 1-substituted 3-phenylprop-2-en-1-one **8** (2.5 mmol) in ethanol (8 mL) was added a corresponding *sec*-amine (2 mL), and the resulting solution was stirred under reflux for 48 h. The solvent was removed under reduced pressure, and the residue was dissolved in CH₂Cl₂ (10 mL), washed with 10% NaOH (2 × 5 mL) and brine, and dried over anhyd Na₂SO₄, and the solvent was removed *in vacuo*. Column chromatography of the crude product (silica gel, eluent hexane: ethyl acetate 5:1) gave the corresponding pyridine **14a–k**. The properties of the prepared pyridines are shown in Table 2.

2-(Diethylamino)-4,6-diphenylpyridine (14a): ¹H NMR δ 1.26 (t, J = 7.1 Hz, 6H), 3.65 (q, J = 7.1 Hz, 4H), 6.58 (s, 1H), 7.20 (s, 1H), 7.34–7.49 (m, 6H), 7.64 (d, J = 6.7 Hz, 2H), 8.10 (d, J = 7.0 Hz, 2H); ¹³C NMR δ 13.2, 42.7, 102.2, 106.7, 126.8, 127.1, 128.3, 128.4, 128.8, 140.4, 140.6, 150.6, 155.7, 157.8.

2-(Pyrrolidin-1-yl)-4,6-diphenylpyridine (14b): ¹H NMR δ 1.99–2.03 (m, 4H), 3.57–3.62 (m, 4H), 6.47 (d, J = 1.1 Hz, 1H), 7.23 (d, J = 1.1 Hz, 1H), 7.33–7.48 (m, 6H), 7.66 (d, J = 6.6 Hz, 2H), 8.10 (d, J = 7.3 Hz, 2H); ¹³C NMR δ 25.6, 47.8, 103.0, 106.9, 126.9, 127.1, 128.3, 128.4, 128.8, 140.3, 150.3, 155.9, 157.7.

2-(Piperidin-1-yl)-4,6-diphenylpyridine (14c): ¹H NMR δ 1.62–1.80 (m, 6H), 3.67–3.69 (m, 4H), 6.77 (d, J = 1.0 Hz, 1H), 7.26 (d, J = 1.0 Hz, 1H), 7.34–7.48 (m, 6H), 7.64 (d, J = 6.6 Hz, 2H), 8.07 (d, J = 7.0 Hz, 2H); ¹³C NMR δ 24.9, 25.6, 46.4, 103.7, 108.1, 126.9, 127.1, 128.4, 128.5, 128.8, 140.2, 140.3, 150.9, 155.7, 159.9.

2-(Morpholin-1-yl)-4,6-diphenylpyridine (14d): ¹H NMR δ 3.63–3.67 (m, 4H), 3.84–3.87 (m, 4H), 6.74 (s, 1H), 7.35 (s, 1H), 7.37–7.48 (m, 6H), 7.63 (d, J = 6.6 Hz, 2H), 8.06 (d, J = 7.0 Hz, 2H); ¹³C NMR δ 45.7, 66.8, 103.5, 109.4, 126.8, 127.1, 128.4, 128.6, 128.7, 128.8, 139.8, 139.9, 151.1, 155.8, 159.7.

2-(Diethylamino)-4-phenyl-6-(4-tolyl)pyridine (14e): ¹H NMR δ 1.58 (t, J = 7.1 Hz, 6H), 2.71 (s, 3H), 4.00 (q, J = 7.1 Hz, 4H), 6.88 (s, 1H), 7.50 (s, 1H), 7.56 (d, J = 8.3 Hz, 2H), 7.71–7.81 (m, 3H), 7.96 (d, J = 7.1 Hz, 2H), 8.26 (d, J = 8.0 Hz, 2H); ¹³C NMR δ 13.2, 21.2, 42.7, 101.9, 106.3, 126.7, 127.1, 128.2, 128.7, 129.1, 137.7, 138.2, 140.7, 150.6, 155.7, 157.7.

2-(Pyrrolidin-1-yl)-4-phenyl-6-(4-tolyl)pyridine (14f): ¹H NMR δ 1.99–2.04 (m, 4H), 2.39 (s, 3H), 3.57–3.62 (m, 4H), 6.46 (s, 1H), 7.20 (s, 1H), 7.24 (d, J = 8.0 Hz, 2H), 7.36–7.48 (m, 3H), 7.66 (d, J = 6.8 Hz, 2H), 8.00 (d, J = 8.0 Hz, 2H); ¹³C NMR δ 21.3, 25.6, 46.8, 102.7, 106.6, 126.8, 127.1, 128.3, 128.7, 129.1, 137.6, 138.2, 140.4, 150.3, 155.9, 157.7. **2-(Piperidin-1-yl)-4-phenyl-6-(4-tolyl)pyridine (14g):** ¹H NMR δ 1.67 (m, 6H), 2.38 (s, 3H), 3.67–3.68 (m, 4H), 6.75 (s, 1H), 7.20–7.25 (m, 3H), 7.38–7.47 (m, 3H), 7.63 (d, J = 6.7 Hz, 2H), 7.97 (d, J = 8.2 Hz, 2H); ¹³C NMR δ 21.2, 24.9, 25.6, 46.4, 103.4, 107.8, 126.8, 127.1, 128.3, 128.8, 129.1, 137.5, 138.3, 140.4, 150.8, 155.7, 159.9.

2-(Morpholin-1-yl)-4-phenyl-6-(4-tolyl)pyridine (14h): ¹H NMR δ 2.42 (s, 3H), 3.69–3.70 (m, 4H), 3.90–3.91 (m, 4H), 6.75 (s, 1H), 7.29 (d, J = 6.8 Hz, 2H), 7.36 (s, 1H), 7.47–7.50 (m, 3H), 7.66 (d, J = 6.9 Hz, 2H), 7.99 (d, J = 7.0 Hz, 2H); ¹³C NMR δ 21.3, 45.8, 66.9, 103.2, 109.1, 126.8,127.1, 128.6, 128.8, 129.2, 137.0, 138.6, 140.0, 151.1, 155.8, 159.7.

2-(Diethylamino)-4-phenyl-6-methylpyridine (14i): ¹H NMR δ 1.19 (t, J = 7.1 Hz, 6H), 2.43 (s, 3H), 3.56 (q, J = 7.1 Hz, 4H), 6.43 (s, 1H), 6.58 (s, 1H), 7.32–7.43 (m, 3H), 7.56–7.58 (m, 2H); ¹³C NMR δ 13.0, 24.8, 42.3, 100.3, 109.1, 127.0, 128.1, 128.6, 140.5, 150.0, 157.2, 157.8.

2-(Piperidin-1-yl)-4-phenyl-6-methylpyridine (14j): ¹H NMR δ 1.64–1.67 (m, 6H), 2.45 (s, 3H), 3.56–3.57 (m, 4H), 6.62 (s, 1H), 6.66 (s, 1H), 7.34–7.45 (m, 3H), 7.55–7.59 (m, 2H); ¹³C NMR δ 24.7, 24.8, 25.6, 46.5, 102.0, 110.7, 127.0, 128.2, 128.7, 140.2, 150.3, 157.1, 160.1.

2-(Morpholin-1-yl)-4-phenyl-6-methylpyridine (14k): ¹H NMR δ 2.47 (s, 3H), 3.56 (t, J = 4.9 Hz, 4H), 3.85 (t, J = 4.8 Hz, 4H), 6.61 (s, 1H), 6.77 (s, 1H), 7.35–7.49 (m, 3H), 7.58 (d, J = 9.0 Hz, 2H); ¹³C NMR δ 24.7, 46.0, 66.9, 101.9, 112.2, 127.1, 128.5, 128.8, 139.9, 150.7, 157.3, 160.0.

Reaction of 2-(Benzotriazol-1-yl)acetonitrile (7a) with 1,3-Diphenylprop-2-en-1-one (8a) in the Presence of NaOH. Compounds 7a (0.26 g, 1.8 mmol) and 8a (0.37 g, 1.8 mmol) were added at rt to a stirred solution of NaOH (0.36 g, 9.0 mmol) in ethanol (5 mL). After 18 h, the reaction mixture was diluted with water (20 mL), and the formed precipitate A was filtered, washed with water, and air dried. To the filtrate was added a solution of HCl (4 N, 20 mL), the resulting mixture was stirred for 2 h, and the precipitate B was filtered, washed with water, and air dried. Both crude products were subjected to column chromatography: product A on silica gel using hexanes: CH₂Cl₂ 1:1 as eluting system afforded 0.073 g (15%) of the 2-ethoxy-4,6-diphenylpyridine (13a); product B on silica gel using MeOH:CHCl₃ 1:20 as eluting system afforded 0.210 g (47%) of the 4,6-diphenylpyrid-2-one (15a), mp 216-218 °C (ethyl acetate). Spectral characteristics and mp of 15a are consistent with those determined for another sample of this product obtained by the procedure described below.

2-Ethoxy-4,6-diphenylpyridine (13a): prisms; mp 56–57 °C; ¹H NMR δ 1.45 (t, J = 7.1 Hz, 3H), 4.53 (q, J = 7.1 Hz, 2H), 6.86 (d, J = 1.2 Hz, 1H), 7.37–7.48 (m, 6H), 7.53 (d, J = 1.2 Hz, 1H), 7.63 (d, J = 6.6 Hz, 2H), 8.07 (d, J = 6.9 Hz, 2H); ¹³C NMR δ 14.8, 61.7, 107.1, 111.6, 126.8, 127.0, 128.6, 128.8, 128.9, 138.8, 139.3, 151.9, 155.1, 164.2. Anal. Calcd for

 $C_{19}H_{17}NO:\ C,\ 82.88;\ H,\ 6.22;\ N,\ 5.09.$ Found: C, 82.95; H, 6.41; N, 4.96.

General Procedure for the Preparation of 3,4,6-Substituted Pyrid-2-ones 15a-h. A mixture of the corresponding benzotriazolyl derivative of acet- or propionamide 7b,c (0.46 mmol), 1,3-diarylprop-3-en-2-one 8a-g (0.46 mmol), and NaOH (43 mg, 1.08 mmol) in ethanol (10 mL) was stirred at reflux for 6 h. On cooling, water (15 mL) and HCl (4 N, 15 mL) were added to the mixture. The precipitated crude product was filtered, washed with water, and air dried. Trituration with an appropriate solvent (see Table 3) gave the corresponding pyrid-2-one 15a-h as a crystalline solid. The properties of the obtained pyrid-2-ones are given in Table 3.

4,6-Diphenylpyrid-2-one (15a): ¹H NMR δ 6.75–6.77 (m, 2H), 7.43–7.55 (m, 6H), 7.63–7.66 (m, 2H), 7.80–7.83 (m, 2H), 12.67 (br s, 1H); ¹³C NMR δ 104.8, 115.3, 126.9, 128.9, 129.1, 129.4, 130.0, 133.8, 138.2, 146.9, 153.7, 165.6.

4-Phenyl-6-(4-methylphenyl)pyrid-2-one (15b): ¹H NMR δ 2.41 (s, 3H), 6.72 (d, J = 1.6 Hz, 1H), 6.74 (d, J = 1.6 Hz, 1H), 7.31–7.33 (m, 2H), 7.43–7.50 (m, 3H), 7.62–7.71 (m, 4H), 12.45 (br s, 1H); ¹³C NMR δ 21.3, 104.3, 114.9, 126.7, 126.9, 128.9, 129.4, 129.9, 130.9, 138.3, 140.3, 146.9, 153.7, 165.5.

4-(4-Methylphenyl)-6-(4-methoxyphenyl)pyrid-2-one (**15c):** ¹H NMR δ 2.42 (s, 3H), 3.88 (s, 3H), 6.67 (s, 1H), 6.69 (s, 1H), 7.04 (d, J = 9.0 Hz, 2H), 7.28 (d, J = 9.9 Hz, 2H), 7.54 (d, J = 8.1 Hz, 2H), 7.70 (d, J = 8.7 Hz, 2H); ¹³C NMR δ 21.4, 55.7, 108.9, 111.6, 115.4, 123.6, 127.3, 128.8, 130.4, 132.6, 142.5, 148.5, 159.6, 162.7.

4-(4-Nitrophenyl)-6-phenylpyrid-2-one (15d): ¹H NMR δ 6.80 (s, 1H), 7.12 (s, 1H), 7.46–7.56 (m, 3H), 7.92–7.94 (m, 2H), 8.12 (d, J = 8.5 Hz, 2H), 8.32 (d, J = 8.5 Hz), 12.0 (br s,

1H); ^{13}C NMR δ 111.4, 124.8, 127.1, 128.6, 130.0, 130.9, 132.3, 142.1, 149.2, 149.4, 156.6, 163.7.

4-(4-Chlorophenyl)-6-phenylpyrid-2-one (15e): ¹H NMR δ 6.68 (s, 1H), 7.02 (s, 1H), 7.46–7.53 (m, 3H), 7.56 (d, J=8.3 Hz, 2H), 7.86 (d, J=8.5 Hz, 2H), 7.90–7.94 (m, 2H), 11.7 (br s, 1H); ¹³C NMR δ 111.0, 111.6, 127.2, 128.6, 129.9, 130.0, 131.2, 132.0, 134.0, 138.1, 148.9, 158.0, 163.3.

4,6-Bis(4-chlorophenyl)pyrid-2-one (15f): ¹H NMR δ 6.73 (s, 1H), 7.15 (s, 1H), 7.56 (d, J = 8.3 Hz, 4H), 7.87 (d, J = 8.5 Hz, 2H), 7.97 (d, J = 8.3 Hz), 11.6 (br s, 1H); ¹³C NMR δ 111.3, 111.5, 128.5, 128.6, 129.6, 130.0, 130.2, 133.9, 138.2, 138.7, 147.6, 158.0, 163.5.

4-(3-Nitrophenyl)-6-(naphth-1-yl)pyrid-2-one (15g): ¹H NMR δ 6.66 (s, 1H), 6.74 (s, 1H), 7.48–7.72 (m, 5H), 7.84–8.04 (m, 4H), 8.30 (d, J = 8.1 Hz, 1H), 8.46 (s, 1H), 12.50 (br s, 1H); ¹³C NMR δ 107.0, 116.4, 121.9, 124.1, 124.5, 125.2, 126.6, 127.4, 127.5, 128.8, 130.1, 130.7, 131.6, 132.7, 133.9, 139.7, 146.9, 148.8, 150.9, 164.3.

3-Methyl-4,6-diphenylpyrid-2-one (15h): ¹H NMR δ 2.13 (s, 3H), 6.50 (s, 1H), 7.25–7.49 (m, 8H), 7.80–7.82 (m, 2H), 12.30 (br s, 1H); ¹³C NMR δ 13.5, 107.2, 124.4, 126.5, 128.0, 128.2, 128.4, 129.0, 129.6, 133.6, 139.9, 142.4, 151.4, 165.4.

Supporting Information Available: ¹H and ¹³H NMR spectra for compounds **14i** and **15d,g** (9 pages). This material is contained in libraries on microfiche, immediately follows this article in the microfilm version of the journal, and can be ordered from the ACS: see any current masthead page for ordering information.

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